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experience and  
exploit by spending  
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when? pull off you

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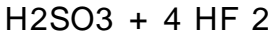
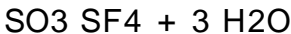
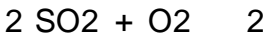
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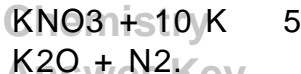
in this set (32)

Which one of the  
following reactions  
is NOT balanced? 2





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ANSWER: Correct

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Question 2 Part A

Iron reacts with  
oxygen in the air to  
produce rust

( $\text{Fe}_2\text{O}_3$ ) according

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to the reaction  
shown below.  $4 \text{ Fe}$   
 $+ 3 \text{ O}_2 \rightarrow 2 \text{ Fe}_2\text{O}_3$   
How many moles of  
rust can be formed  
if 1.8 mols of iron  
are present? Hint 1.  
Begin with your  
given number of  
moles.

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8 CHAPTER 8

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Answer the

following questions

in the space

provided. 1. List

four metals that will

not replace

hydrogen in an acid.

Choose from Cu,

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Ag, Au, Pt, Sb, Bi,  
and Hg.

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8 Answer Key

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2: Rank from most  
to fewest valence  
electrons of the  
following elements:  
Kr, Ga, Br, As, K. -  
Answer: Kr (8), Br

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(7), As (5), Ga (3),

K (1). Valence

electrons are equal  
to the atom's main

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tools. On the  
periodic table, one  
of the columns  
within the main  
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similar chemical  
properties. (3.2). ht  
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