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Section 2

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Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

mc06se cFMsr i-vi

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Solutions SECTION 2 SHORT
ANSWER Answer the following
questions in the space
provided. 1. The following

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Answers are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in ...

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Chemistry. solution
equilibrium. saturated
solution. unsaturated
solution. supersaturated
solution. physical state in
which the opposing processes
of dissolution... solution
that contains the maximum
amount of dissolved solute.
solution that contains less
solute than a saturated
solution u...

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CHAPTER 12 REVIEW Solutions
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Answer the following
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Chapter 2 of NCERT Chemistry
is based on the basic idea
of Solutions and its types.
This chapter covers
Concentration of Solutions,
Solubility, Vapour Pressure
of Liquid Solutions, Ideal
and Non-Ideal Solutions,
Colligative Properties and

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Determination and Abnormal
Molar Masses.

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QUESTIONS SOLVED 2.1.

Calculate the mass
percentage of benzene (C_6H_6)
and carbon tetrachloride
(CCl_4) if 22 g of benzene is
dissolved in 122 g of carbon
tetrachloride.

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alcohol b. water c. the gels
3.

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